

AMINATED SURFACE

TECHNICAL NOTE N. 12

General directions for the use of surfaces with amino and carboxylic groups

Biomat has developed modified polystyrene surfaces introducing chemical groups such as NH₂ and COOH.

These groups are able to covalently bind compounds to the plastic surface. The optical properties of polystyrene remain unchanged, allowing to use the modified surfaces as powerful tools for diagnostic assays.

These surfaces offer the possibility to

- · covalently immobilize small molecules which only bind weakly or not at all by physical adsorption
- orientate the immobilization of molecules in a defined way on the solid phase.

Hereunder are some examples of application that can be used as guidelines to enable users to develop their own biospecific assays.

1. Coupling of NHS-activated compounds

A very simple and easy application of Biomat NH_2 surfaces is coupling of molecules that have been activated by esterification with N-hydroxysuccinimide derivatives (NHS). In our experiment an-Hydroxysuccinimide active ester of biotin links immediately via its carbonilic group to the surface amino groups as shown in figure 1.

Figure 1

Biotin coupled to Biomat NH₂ surface

Preparation of reagents and buffers

Materials

Solid phase:	Biomat plates	MT02F2-AM1 (primary amino groups) MG01F-HB (high binding capacity)
ε-Caproylamido-biotin-N- hydroxysuccinimide ester (NHS- biotin)	BIO-SPA	Cat No. B002-61
Dimetilformamide (DMFO)	Fluka	Cat No. 40250
Tween® 20	Merck	Cat No. 822184
Streptavidin	BIO-SPA	Cat. No. S002-60
Streptavidin-peroxidase conjugate	BIO-SPA	Cat. No. SB01-61
BSA	Intergen	Cat. No. 3100
TMB peroxidase substrate	Kirkegard & Perry	Cat. No. 50-76-05



NHS-Biotin stock solution

NHS-biotin 6mg DMFO 2 ml

NHS-Biotin solution 150µg/ml

NHS Biotin stock solution $$500\mu l$$ PBS 0.1M pH 7.2+0.15% Tween $^{\circledR}$ 20 $\rightarrow \! 10 ml$

NHS-Biotin solution 100µg/ml

NHS Biotin stock solution $$333\mu l$$ PBS 0.1M pH 7.2+0.15% Tween $^{\circledR}$ 20 $\rightarrow \! 10 ml$

NHS-Biotin solution 50µg/ml

NHS Biotin stock solution $$167\mu l$$ PBS 0.1M pH 7.2+0.15% Tween $^{\circledR}$ 20 $\rightarrow \! 10ml$

NHS-Biotin solution 10μg/ml

NHS Biotin stock solution $$33\mu l$$ PBS 0.1M pH 7.2+0.15% Tween $^{\circledR}$ 20 $\rightarrow \! 10m l$

Streptavidin-mix

Streptavidin	50μg
Streptavidin-peroxidase	1μg
PBS-BSA 1%	10ml

Experiment

- 1. Add 100μ l NHS-biotin solutions 150-100-50-10 μ g/ml and 0.1M PBS + Tween $^{\circledR}$ 20 0.15% pH 7.2 as 0 μ g/ml to the wells (with primary amines and HB). Seal the wells with adhesive tape to prevent evaporation
- 2. Incubate overnight at room temperature
- 3. Empty the wells and wash with 0.1M PBS + Tween® 20 0.05%, pH 7.2 four times
- 4. Add 100 µl of streptavidin mix to each well and incubate 30 minutes at room temperature
- 5. Empty the wells and wash with 0.1M PBS + Tween[®] 20 0.05%, pH 7.2 four times
- 6. Add 100μ /well of TMB substrate solution and incubate 10 minutes at room temperature
- 7. Stop the substrate reaction by adding 100 μ l of sulphuric acid 1 N and read the optical density values at 450 nm

Results

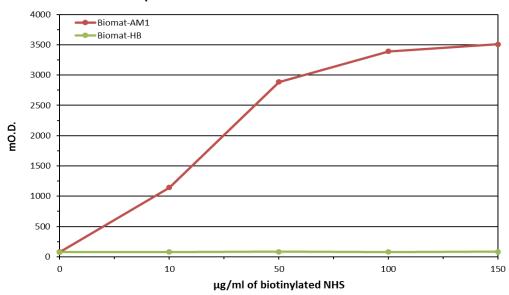
The results (see fig. 2) show a clear correlation between the concentration of NHS-biotin added to the wells and the amount of biotin bound to the Biomat NH_2 surface.

On the other side no biotin is bound onto the plate without primary amino groups grafted to its surface, showing that passive adsorption of neither biotin nor enzyme conjugate occurs.

We therefore conclude that NHS-Biotin has indeed been covalently bound to the amino groups present on the Biomat NH_2 surface.

Figure 2

Comparison between Biomat-AM1 and Biomat-HB



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2. Coupling hapten or peptide, having a carboxylic group, to Biomat NH₂ surface

The carboxylic group presents in a molecule with a low molecular weight, such as a hapten or a peptide, binds to Biomat NH₂ through formation of amide bonds between the carboxylic group presents in the molecule and the surface amino group by the combined action of carbodiimide and N-hydroxysuccinimide.

The figure 3 shows the reaction scheme for coupling of the hapten, biotin, through its available carboxylic group.

$$R_{1} = CH_{2}-CH_{3}$$

$$R_{2} = (CH_{2})_{3}-N-CH_{3}$$

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$$R_{2} = (CH_{2})_{3}-N-CH_{3}$$

$$R_{2} = (CH_{2})_{3}-N-CH_{3}$$

$$R_{3} = CH_{2}-CH_{3}$$

$$R_{2} = (CH_{3})_{3}-N-CH_{3}$$

$$R_{3} = CH_{3}-CH_{3}$$

$$R_{4} = CH_{4}-CH_{3}$$

$$R_{5} = CH_{5}-CH_{5}$$

$$R_{5} = CH_{$$

Biotin coupled to Biomat NH2 surface

NHS



Preparation of reagents and buffers

Materials

Solid phase:	Biomat plates	MG02F-AM1(primary amino groups) MG01F-HB (high binding capacity)
d-Biotin	Sigma	Cat. No. B 4501
1-Ethyl-3-(3 dimethylaminopropyl)-carbodiimide (EDC)	Sigma	Cat. No. E 1769
Sulfo-N-hydroxysuccinimide (sulfo-NHS)	Fluka	Cat. No. 56485
Dimethylsulfoxide (DMSO)	Merck	Cat. No. 2931
Tween® 20	Merck	Cat. No. 822184
Streptavidin	BIO-SPA	Cat. No. S002-60
Streptavidin-peroxidase conjugate	BIO-SPA	Cat. No. SB01-61
BSA	Intergen	Cat. No. 3100
TMB peroxidase substrate	Kirkegard & Perry	Cat. No. 50-76-05

Biotin stock solution	EDC solution
BIOTIN STOCK SOLUTION	EDC SOlution

d-Biotin	7.8 mg	EDC	5.8 mg
DMSO	0.5 ml	Distilled water	to 10 ml
Distilled water	0.5 ml		

Biotin/NHS solution Streptavidin-mix

Biotin stock solution	500µl	Streptavidin	50 μg
Sulfo-NHS	3.45 mg	Streptavidin-peroxidase	1μg
Distilled water+0,30% Tween®	20 to 10 ml	PBS-BSA 1%	10ml

Experiment

- 1. Add 50 μ l of distilled water to each well, apart from wells in column 2. Then add 100 μ l of Biotin-NHS solution to all wells in column 2
- 2. Dilute by transferring 50 μ l from the wells in column 2 to column 3, mix, transfer 50 μ l from column 3 to column 4, mix and proceed in this way up to column 12
- 3. To start reaction: add 50 μ l of EDC solution to each column. In blank experiment (column 1) add 50 μ l of distilled water instead of EDC
- 4. Incubate at room temperature for 2 hours
- 5. Empty the wells and wash with 0.1M PBS+0.05% Tween[®] 20 pH 7.2 four times
- 6. Add $100\mu l$ of streptavidin mix to each well and incubate for 30 minutes at room temperature
- 7. Empty the wells and wash with 0.1M PBS+0.05% Tween $^{\circledR}$ 20 pH 7.2 four times
- 8. Add 100 μ l of TMB substrate solution to each well and incubate for 10 minutes at room temperature
- 9. Stop the substrate reaction by adding 100 μ l of sulphuric acid 1 N and read the optical density values at 450 nm

Results

The results of this experiment (figure 4) clearly show that the molecule (biotin) is bound in a detectable way to the Biomat NH_2 (cod. AM1) whereas no detection could be obtained on Biomat HB.

The results indicate that a covalent coupling has taken place between the carboxylic group in the biotin and the primary amino group grafted on the Biomat NH_2 . The results (data not displayed) point out that without adding carbodiimide the covalent binding of biotin does not occur.

Figure 4

